Translational Energy Distribution from $C_2H_2 + h\nu(193.3 \text{ nm}) \rightarrow C_2H + H^+$

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We report the center-of-mass translational energy distribution for 193.3-nm photolysis of expansion-cooled $C_2H_2$, using high-$n$ Rydberg-level excitation of nascent H atoms to facilitate high-resolution time-of-flight measurements. The observed resolution of $\sim 200$ cm$^{-1}$ is presently limited by the ArF photolysis laser, whose band width is approximately 200 cm$^{-1}$. The reported distribution clearly resolves a $C_2H$ bending progression ($v_3 \sim 400$ cm$^{-1}$), which reflects the trans-bent $C_2H_2$ excited state. A definitive assignment of all features is still not possible on the basis of currently available information on $C_2H$. These measurements yield an upper bound to $D_0$ of 131.8 $\pm 0.5$ kcal mol$^{-1}$.

Introduction

The spectroscopy and photochemistry of acetylene have been studied extensively, both experimentally$^1$-$^8$ and theoretically.$^9$ Early spectroscopic investigations showed a prominent bending progression in the $A''_1 \leftarrow X''_2$,$^{10}$ absorption spectrum, indicating that the first electronically excited state is trans bent.$^{11,12}$ Photochemical studies showed that photolysis at wavelengths shorter than $\sim 200$ nm yields the ethynyl radical, $C_2H$, and an H atom.$^3$ Wodtke and Lee reported the center-of-mass (CM) translational energy distribution following 193.3-nm photolysis using the angle-resolved time-of-flight (TOF) method, and gave an upper

\begin{enumerate}
  \item Innes, K. K. J. Chem. Phys. 1955, 22, 863.
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\end{enumerate}

\footnotesize

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\end{itemize}
Photolysis of Expansion-Cooled C$_2$H$_2$

Figure 1. Excitation scheme for the detection of H atoms via high-lying Rydberg levels.

bound to the dissociation energy of $D_0$(C$_2$H--H) = 132 ± 2 kcal mol$^{-1}$. Yet important questions remain. In particular, $D_0$(C$_2$H--H) has remained the focus of experimental and theoretical study, and the fragmentation dynamics of the excited C$_2$H$_2$ reactive precursor are essentially unknown. Most recent studies have yielded upper limits for $D_0$ around 131 kcal mol$^{-1}$, while values around 127 kcal mol$^{-1}$ have also been reported.

Likewise, the ethynyl radical has also been a fertile area for study. C$_2$H has been observed directly, although Hirota and co-workers derive a soot. It has been investigated by ESR, infrared and visible cm$^{-1}$ above the ground state, assignment of the C$_2$H remaining elusive. While both theory and experiment play an important role in combustion and in the formation of low-temperature matrices by Jacox and in the gas phase at 1841 cm$^{-1}$ from C$_2$H created by 193-nm C$_2$H$_2$ photolysis.

Figure 2. (a) H-atom TOF spectra, obtained using a 23.6-cm flight path (offset by an arbitrary amount). (b) Data from the upper TOF spectrum transformed to a CM translational energy scale.

In this article, we report the C$_2$H$_2 + h\nu \rightarrow$ H + C$_2$H center-of-mass translational energy distribution for 193.3-nm photolysis, at higher resolution (i.e., ~200 cm$^{-1}$) than has been reported previously. The present results are the first to clearly resolve the $v_2$ C$_2$H bending progression that dominates the distribution. The Rydberg TOF method employed offers the possibility of high translational energy resolution (i.e., tens of cm$^{-1}$), which may ultimately be needed to establish $D_0$ beyond reasonable doubt. Although we cannot yet assign all of the observed spectral features, the present results offer insight into the photophysics of C$_2$H$_2$ photodissociation, as well as the physical properties of the ethynyl radical.

Experimental Section

The technique of detecting H atoms via 2-photon, 2-frequency excitation to a high-lying Rydberg state followed by efficient field ionization at the desired detection region was proposed by Letokhov and developed by Welge and co-workers. The principles are shown schematically in Figure 1. Briefly, a supersonic expansion of C$_2$H$_2$ seeded in He is skimmed and crossed with a portion of the output from an ArF excimer laser, which is weakly focused by using a 1-m lens. H-atom fragments are excited by VUV light tuned to the line center of the Lyman $\alpha$ transition at 121.6 nm. The Lyman $\alpha$ radiation is generated by tripling the output of an excimer laser-pumped dye laser (364.7 nm) in Kr.

Another excimer laser-pumped dye laser beam, overlapping the 121.6-nm beam, transfers H-atom population from the 2$P$ level to a high-lying Rydberg level (e.g., n = 40–90). In contrast to the rapidly fluorescing 2$P$ state, high-n Rydberg levels are radiatively metastable and remain highly excited for many tens of microseconds. These excited atoms travel to a detector 23.6 cm from the interaction region, with Doppler selection in the Lyman $\alpha$ step.


ensuring that the largest possible fraction of excited H atoms are sent toward the detector. Since the excited atoms are neutral, they are not affected by stray fields and space charge. The latter is a major problem when detecting proton TOF following laser ionization in the photodissociation region.25 Once the excited H atoms arrive at the detector, they are easily field-ionized and are then detected as ions. TOF spectra are recorded by using a transient digitizer and a computer.

Results and Discussion

Figure 2a shows two TOF spectra. Although relative intensities differ slightly for the two traces, the main features are very reproducible: a small unstructured contribution at 15–20 μs, followed by a series of well-resolved peaks. The 15–20-μs feature will be discussed below. Figure 2b shows the structured region of the upper TOF spectrum transformed (including Jacobian) to a CM translational energy scale. Contributions from the beam velocity are minimal due to high C2H2/He ratios and correspondingly low beam velocities (~10^8 cm s^-1). Since the CM translational energy distribution mirrors the C2H internal energy distribution, it is evident that C2H is highly excited. These results agree qualitatively with the distribution reported by Wodtke and Lee,4 but the present measurements reveal considerably more detail. In particular, there is a striking progression of features between 3000 and 5500 cm^-1, spaced ~410 cm^-1 apart. Since only the νb bending mode of C2H has a frequency near this value, these peaks clearly form a bending progression. This is not surprising, since the equilibrium geometry of the predissociating C2H2 A state is trans bent, whereas the equilibrium geometry of C2H is linear in both the X2Σ and A2Π states. Between 2000 and 3000 cm^-1, there are three closely spaced, yet resolved, peaks, followed by the two most prominent features in the spectrum. No signal is observed below ~1000 cm^-1.

Our tentative assignment for the 15–20-μs signal is C2H photolysis:

\[ \text{C}_2\text{H} + h\nu(193.3 \text{ nm}) \rightarrow \text{C}_2 + \text{H} \]  

(1)

This is known to occur in 193.3-nm C2H2 photolysis and has been studied recently by Jackson and co-workers.29 Although this could, in principle, result in a structured spectrum, it seems likely that several different C2H vibrational levels will absorb photolysis radiation and dissociate, resulting in a relatively structureless spectrum. It is possible that this contribution to the TOF spectrum is responsible for the signal observed in the wings of Doppler spectra.7 Preliminary results indicate that the 15–20-μs signal has a similar photolysis fluence dependence to that of the main signal. This is agreement with that reported in the Doppler work; this is consistent with the C2H absorption cross section being much higher than that of C2H2 photolysis. If C2H2 photolysis is responsible for the 15–20-μs signal, it is likely that the signal persists at longer arrival times and the structured part of the TOF spectrum from 20 to 40 μs lies on top of an unstructured background. Another explanation is that the 10–15-μs signal results from a small number of highly excited hydrogen atoms that are field ionized in the TOF tube.

To better understand the results, we simulated the observed distribution by convoluting a "stick spectrum" with the measured experimental resolution. Photodissociation experiments with HBr and HI at 193 and 248 nm showed that the C2H2 experimental resolution is given by the band width of the ArF photolysis laser, \( \Delta E_{\text{FWHM}} \approx 200 \text{ cm}^{-1} \). The feature in the experimental spectrum around 3000 cm^-1 appears to be a doublet, and is thus simulated by using two lines. The other peaks are better resolved. The stick spectrum and its convolution that best fits the data are shown in Figure 3a, and the results are summarized in Table I. For purposes of discussion, the 13 resolved features are labelled a–m. Peaks g–k and peak m are well reproduced in the simulation, but the simulated peaks i and a–f are narrower than the data. Closer inspection of the data shows that peaks a–f appear to degrade toward lower energies, suggesting rotational excitation. Consequently, we also simulated the data using rotational distributions for the a–f peaks, as shown in Figure 3b and listed in Table II. The distributions were chosen ad hoc, but some degree of rotational excitation was necessary in order to obtain the improved fit of Figure 3b relative to 3a.

The relative energies of the "origins" (i.e., \( J = 0 \) levels listed in Table II) used to simulate the a–f features can be fit by using a νb frequency of 420 cm^-1, as shown in Figure 4. Given the present low resolution, anharmonicity is ignored. The corresponding plot of "stick spectrum" values from Table I is similar, with a slightly lower frequency, 410 cm^-1. The apparent g/h doublet ~3000 cm^-1 can be included as the sixth member of the progression by averaging the frequencies of the sticks used to simulate the two peaks. Interestingly, if one assumes that l and

\( \Delta E_{\text{FWHM}} \approx 200 \text{ cm}^{-1} \)

(2)

Estimated absolute uncertainty ±0.3 kcal mol^-1 (±100 cm^-1).

Estimated relative uncertainty ±50 cm^-1.
TABLE II: Parameters for Spectrum Associated with Figure 3b

<table>
<thead>
<tr>
<th>Progression Member</th>
<th>CM Translational Energya (kcal mol⁻¹ cm⁻¹)</th>
<th>Rel. Energy Below Peak a (cm⁻¹)b¢</th>
<th>Intensity</th>
</tr>
</thead>
<tbody>
<tr>
<td>a</td>
<td>16.10</td>
<td>5630</td>
<td>0.0436</td>
</tr>
<tr>
<td>b</td>
<td>15.00</td>
<td>5250</td>
<td>0.0106</td>
</tr>
<tr>
<td>c</td>
<td>13.90</td>
<td>4860</td>
<td>0.0132</td>
</tr>
<tr>
<td>d</td>
<td>12.75</td>
<td>4440</td>
<td>0.0140</td>
</tr>
<tr>
<td>e</td>
<td>11.75</td>
<td>3950</td>
<td>0.0145</td>
</tr>
<tr>
<td>f</td>
<td>10.15</td>
<td>3530</td>
<td>0.0186</td>
</tr>
<tr>
<td>g</td>
<td>9.00</td>
<td>3150</td>
<td>0.23</td>
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<tr>
<td>h</td>
<td>8.50</td>
<td>2970</td>
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<tr>
<td>i</td>
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<td>0.34</td>
</tr>
<tr>
<td>j</td>
<td>7.10</td>
<td>2480</td>
<td>0.34</td>
</tr>
<tr>
<td>k</td>
<td>6.40</td>
<td>2240</td>
<td>0.40</td>
</tr>
<tr>
<td>l</td>
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<td>0.68</td>
</tr>
<tr>
<td>m</td>
<td>4.10</td>
<td>1430</td>
<td>1.00</td>
</tr>
</tbody>
</table>

*a Estimated absolute uncertainty ±0.3 kcal mol⁻¹ (±100 cm⁻¹).
*b Estimated relative uncertainty ±50 cm⁻¹. 'Value for the J = 0 member of the Boltzmann distribution; see text.

Figure 4. CM translational energy below a (from Table II) vs. progression member: af (origins) are attributed to members 0 to 5, g and h are averaged for 6, while l and m are attributed to 9 and 10. The slope of the straight line is 420 cm⁻¹. Similar results are obtained using energies from Table I.

m are the ninth and tenth members of the progression, they fall on the same straight line as the other points. This suggests that l and m might be attributable to a single progression in the X2Σ state or at least that they are attributable to a bending progression with a similar frequency, i.e., in the A2Π state.

Because of the Π vibronic symmetry of the υ2 bending mode, it is expected that X2Σ states with odd quanta of υ2 will interact strongly with the A2Π state.22 However, there is no evidence of this in the distribution. Similarly, Kanamori and Hirota report observation of a band at 2166 cm⁻¹ that they assign to the X2Σ-(07'0) upper state. This corresponds to a much smaller υ2 spacing than we see here. We note that our results are more consistent with an assignment of the observed 2166-cm⁻¹ band as X2Σ(05'0).

In addition to C2H bending, what can explain the other features in the spectrum? In the g-k region (~2450-3400 cm⁻¹ less translational energy than a), the simple pattern seen in a-f is broken, and there are several closely spaced peaks. Of course, with the present 200-cm⁻¹ resolution, many features may simply be unresolved.

As discussed above, A2Π is thought to be 3500-4000 cm⁻¹ above X2Σ. Also, it is known that A2Π mixes with nearby X2Σ vibrational levels, and this perturbation has made a definitive assignment of spectral features seen in absorption difficult.22 So where is A2Π in Figure 3? One possibility is that a peak corresponds to X2Σ(000), and peak I derives in whole or part from A2Π(000). It lies ~3600 cm⁻¹ above peak a, close to the expected A2Π origin. Also, as noted above, peak I is broader than its neighbors, possibly because of coupling to X2Σ vibrational levels. If this is the case, peak m, which is 470 cm⁻¹ higher in internal energy than 1 and is the dominant feature in the spectrum, can be assigned to A2Π(010), which was seen in emission by Fletcher and Levine.17 The peaks i-k would then correspond to ~2900-3100 cm⁻¹ of C2H internal energy, and therefore one or more of these peaks might be related to υ1 CH stretch, which remains unassigned.

Another possibility is that the messy g-k region corresponds to the perturbed A2Π origin. If this is the case, peak a corresponds to υ2 > 0, and D0 is less than 131.8 kcal mol⁻¹.

From conservation of energy, it is possible, at least in principle, to deduce D0 from knowledge of the CM translational energy distribution:

$$hv - D_0(C_2H-H) = E_{internal}(C_2H) + E_{CM}$$

where it is assumed that C2H2 has negligible internal energy because of expansion cooling. The highest ECM feature that we assign to C2H2 photodissociation appears at 16.1 ± 0.5 kcal mol⁻¹ (5630 ± 170 cm⁻¹), as shown in Table II. If this corresponds to ground vibronic state C2H, then D0 = 131.8 ± 0.5 kcal mol⁻¹. This value is in good agreement with other determinations, both experimental and theoretical.46-8 As mentioned above, it appears that the lower value inferred from Doppler-shift measurements can be explained by C2H photolysis.

This measurement, like others, gives only an upper bound to D0. If the C2H fragment is formed in its ground vibrational level, then the measurement yields D0 directly, but if not, the energy of the lowest occupied level must be subtracted from the upper bound in order to get D0. What is the likelihood of the lowest state being unoccupied? In our opinion, one should not dismiss this possibility too quickly. There is a monotonic increase in population for C2H levels having bending excitation, and there is no a priori reason to believe that peak a corresponds to υ2 = 0. These peaks are part of a progression that gains intensity from a to d, and it is possible that the lowest members of the υ2 progression are simply absent. Thus, the lowest observed level might correspond to υ2 ≥ 1, in which case the 131.8 kcal mol⁻¹ number must be reduced by the energy of the lowest occupied bending level. If this were the case, the A2Π(000) state might lie in the g-k region. Because of this ambiguity, it is not possible to establish D0 solely on the basis of the present measurements. However, with straightforward experimental improvements, we anticipate close to an order of magnitude improvement in resolution.

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